EERC. UND NORTH DAKOTA.

Energy & Environmental Research Center (EERC)

Hydrogen Opportunities in North Dakota

Chad Wocken Assistant Director, Clean Energy Solutions

October 13, 2021

© 2021 University of North Dakota Energy & Environmental Research Center.









CENTER

TECHNOLOGY

CHEMICAL STORAGE

OFFICES

IN-HOUSE FABRICATION SHOP

> // TECHNOLOGY DEMONSTRATION

MOBILE LABORATORIES

OUR FACILITIES 254,000 SQ FT OF FACILITIES

DOE-DESIGNATED NATIONAL CENTER FOR HYDROGEN TECHNOLOGY®

- 15,000-ft² NCHT facility.
- Hydrogen production, separation—purification, storage, and integrated fuel cells testing.
- Special materials lab for development of materials, advanced joining techniques, and application evaluations.
- Fuel cell test facility: flexible test center capable of testing low- and high-temperature fuel cells; customizable fuel delivery system.
- Pilot systems: high-pressure fluidized-bed gasification system; pressurized lab-scale entrained-flow gasifier; hydrogen purification; and other systems for developing hydrogen, alternative fuels, and chemicals.







Station



1-kW Syngas/NG SOFC Stack Test Station

Multicell Test Station

Critical Challenges. Practical Solutions.



WHY HYDROGEN?

- High energy content. On a mass basis, H₂ beats all conventional fuels.
- Numerous feedstocks/production scenarios:
 - Reform natural gas, natural gas liquids/condensates, light oils, other hydrocarbons
 - Gasify coal and/or biomass
 - Electrolyze water using fossil fuel or renewable electricity
- Clean whether combusted or converted to electricity in fuel cell, emission is water: $2H_2 + O_2 \rightarrow 2H_2O$
- In addition to energy/electricity generation, numerous uses include:
 - Petroleum, renewable fuel, and metals refining.
 - Feedstock for production of ammonia, methanol, and other commodity and highervalue products.
- When produced using renewable energy or fossil energy with CO₂ capture, near-zero life cycle carbon emissions.

EERC. UND NORTH DAKOTA

WHY HYDROGEN? ENERGY CONTENT

Mass-based energy content vs. volume-based energy content



1 kg of H_2 contains about the same amount of energy as 1 gallon of gasoline.



1 kg H_2 , compressed to 1000 psig



1 kg H_2 , liquefied

1 kg H_2 , as ammonia





HYDROGEN OPPORTUNITIES IN NORTH DAKOTA

Hydrogen and Power Production

Hydrogen Uses



HYDROGEN OPPORTUNITIES IN NORTH DAKOTA

Hydrogen and Power Production

Hydrogen Uses



HYDROGEN ENERGY ROAD MAP FOR NORTH DAKOTA

State-funded study began in August 2021. Project scope includes:

- Identification and quantification of scale:
 - Magnitude of hydrogen production from North Dakota's resources
 - Possible demand from various sectors, near- and long-term
 - Magnitude of infrastructure needed: storage and transport
 - Impacts of hydrogen use on emissions
- Notional timeline and cost for commercial deployment of various hydrogen energy technologies and approaches.
- Assessment of opportunities to grow North Dakota hydrogen energy economy by leveraging and adding value to agriculture, oil and gas, and electrical generation industries while preserving land, water, and air resources for future generations.



OTHER EEEC HYDROGEN ACTIVITIES

- Hydrogen policy, pricing, and performance modeling how do you incorporate hydrogen into our energy economy?
- Integration with CCS
- Salt cavern storage identification of suitable sites for development
- Electro- and thermochemical process development for hydrogen production and use
- Fuel cell development:
 - Fuel cell material, component, and systems
 - Low- and high-temperature fuel cells
- Vehicle development balance of plant support
- On-demand hydrogen production optimization of previously developed technology



DOE SUPERTRUCK (PROPOSED)

A proposed \$53M project to demonstrate a hydrogen-fueled freight corridor in North Dakota



EERC. UN NORTH DAKOTA.

Chad Wocken Assistant Director cwocken@undeerc.org 701.777.5273 Energy & Environmental Research Center University of North Dakota 15 North 23rd Street, Stop 9018 Grand Forks, ND 58202-9018

www.undeerc.org 701.777.5000 (phone) 701.777.5181 (fax)



The Colors of Hydrogen

 Unofficial definitions have been developed for hydrogen, based on the production method.

Image from:

https://nacfe.org/wpcontent/uploads/2020/12/Hydrogen-Color-Spectrum-HiRes-2.png

Hydrogen Color Spectrum

GREEN

Hydrogen produced by electrolysis of water, using electricity from renewable sources like hydropower, wind, and solar. Zero carbon emissions are produced.

PINK/PURPLE/RED

Hydrogen produced by electrolysis using nuclear power.

YELLOW

Hydrogen produced by electrolysis using grid electricity.

WHITE

Hydrogen produced as a byproduct of industrial processes.

TURQUOISE

Hydrogen produced by the thermal splitting of methane (methane pyrolysis). Instead of CO₂, solid carbon is produced.

BLACK/GRAY

Hydrogen extracted from natural gas using steam-methane reforming.

BLUE

Grey or brown hydrogen with its CO₂ sequestered or repurposed.

BROWN

Hydrogen extracted from fossil fuels, usually coal, using gasification.



Note: There are no official definitions of these colors, but the above represents common industry nomenclature.

Hydrogen Technologies Program



From producing hydrogen molecules through dispensing to end-use applications

